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Secondary metabolites from *Isodon ternifolius* (D. Don) Kudo and their anticancer activity as DNA topoisomerase IB and Tyrosyl-DNA phosphodiesterase 1 inhibitors



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ABSTRACT

Keywords: Isodon ternifolius Secondary metabolite DNA topoisomerase Tyrosyl–DNA phosphodiesterase Cytotoxicity Ce, and **22** acted as TOP1 catalytic inhibitors with equipotent TOP1 inhibition to camptothecin (+ + + +). Compounds **7** and **8** exhibited significant cytotoxicity against MCF-7, A549, and HCT116 cells with GI

 $_{50}$ values in the range of 2.2–4.8 $\mu M.$ This work would provide valuable information that secondary metabolites from *I. ternifolius* could be developed as anticancer agents.

1. Introduction

DNA topoisomerase IB (TOP1) is an essential enzyme that regulates DNA topological structure in many cellular metabolic processes including replication and transcription.¹ TOP1 is a validated molecular target for the discovery of anticancer agent.^{2,3} TOP1 inhibitors are classified into two types as TOP1 "poisons" and "catalytic inhibitors" based on their molecular mechanism of action. TOP1 poisons can stabilize the enzyme-DNA covalent complexes (TOP1cc) to prevent further relegation and lead to DNA single-strand breaks.^{4,5} Camptothecin (CPT) and their derivatives are the classical TOP1 poisons. To date, four wellknown CPT derivatives have been approved for clinical treatment of cancers.^{6–8} Unlike TOP1 poisons, catalytic inhibitors act at the upstream stage of the catalytic DNA cleavage reaction of TOP1, and prevent the formation of TOP1cc by blocking the nucleophilic attack of the scissile strand.⁹ Several catalytic inhibitors from natural products have been already discovered.^{10–12}

Tyrosyl-DNA phosphodiesterase 1 (TDP1) is a DNA repair enzyme that specifically hydrolyzes 3 -phosphotyrosyl bond derived from TOP1cc and repairs TOP1-induced DNA damages.^{13,14} resulting in cancer cells resistant to TOP1 inhibitors.¹⁵ Therefore, inhibition of TDP1 could potentiate the anticancer e cacy of TOP1 inhibitors. Indeed, several previous studies reported that TDP1 inhibitors exhibit synergistic e ect to TOP1 inhibitors.^{16,17} Therefore, TDP1 is a rational target for the development of anticancer agents.^{14,18,19}

Because of the unique biological functions of TOP1 and TDP1, the discovery of their inhibitors has attracted attentions. However, there is only two kind of dual inhibitors reported now.^{17,20} In our study for anticancer agents from natural products, the ethanol extract of the roots of *Isodon ternifolius* (D. Don) Kudo (Labiatae) showed strong TOP1 inhibition at 50 μ g/mL (Fig. S1) and TDP1 inhibitory activity (85% inhibition at 100 μ g/mL), which inspired us to study its secondary metabolites.

I. ternifolius is perennial subshrubs to shrubs widely distributed in the southwestern region of China.²¹ *I. ternifolius* has been commonly used as folk medicine to treat dysenteric enteritis, icterohepatitis, and inflammatory ailments.^{22,23}*I. ternifolius* is also the major ingredient of a Chinese patent medicine "Fufang Sanyexiangchacai Pian", which is

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Fig. 1. Chemical structures of 1-27 isolated from the roots of *I. ternifolius*.

used to treat acute and chronic hepatitis and hepatitis B. Its secondary metabolite oridonin exhibits significant anticancer activity for human breast cancer,²⁴ gallbladder cancer,²⁵ leukemia,²⁶ and gastric cancer.²⁷ In addition, previous phytochemical studies on this plant have revealed that *ent*-kaurane diterpenes, ursanes and oleananes triterpenoids are its main secondary metabolites, some of which show anti-inflammatory,²⁸ antioxidative,²⁹ and anticancer activities.³⁰ In this work, we report the isolation and structure elucidation of the compounds isolated from this

plant, and their biological evaluations.

2. Results and discussion

2.1. Phytochemical investigation

The air-dried powder of the roots of *I. ternifolius* was extracted with 95% EtOH at room temperature to give a crude extract, which was

Table 1 1 H (500 MHz) and 13 C (125 MHz) NMR data of 1 in CDCI_{3.}

No.	_H (mult., J in Hz)	_c , type	No.	_H (mult., J in Hz)	_{c,} type
1	2.12, m	37.6, CH ₂	11	1.74, m	21.9, CH ₂
	1.10, dd (14.0,4.0)			1.74, m	
2	2.69, m	34.6, CH ₂	12	1.86, m	29.1, CH ₂
	2.29, dd (14.7, 3.8)			1.38, m	
3		216.1, C	13	2.97, ddd	41.1, CH
				(10.8, 5.8, 5.8)	
4		47.4, C	14	4.75, d (5.5)	81.9, CH
5	1.70, m	50.4, CH	15		141.7, C
6	2.16, m	24.4, CH ₂	16	6.15, s	120.8, CH ₂
	2.12, m			5.58, s	
7	6.07, m	133.1, CH	17		170.4, C
8		131.6, C	18	1.08, s	25.5, CH ₃
9	2.01, dd (8.7, 5.9)	47.4, CH	19	1.12, s	22.6, CH ₃
10		35.2, C	20	1.01, s	14.2, CH ₃

suspended in H_2O and partitioned with EtOAc. Various column chromatographic separations of the EtOAc fraction a orded twenty seven compounds (Fig. 1), including two new compounds (1 and 13) and twenty five known compounds (2–12 and 14–27).

Isodopene A (1) was obtained as white powder. Its molecular formula $C_{20}H_{26}O_3$ was determined by the positive HRESIMS ion at m/z337.1771 [M + Na]⁺ (calcd for $C_{20}H_{26}O_3Na^+$, 337.1774). The IR spectrum revealed the presence of the carbonyl groups (1762 and 1707 cm⁻¹). The ¹H NMR data of **1** displayed signals for three methyl groups [H 1.01 (3H, s), 1.08 (3H, s) and 1.12 (3H, s)], a terminal double bond [$_{\rm H}$ 5.58 (1H, s) and 6.15 (1H, s)], an olefinic proton [$_{\rm H}$ 6.07 (1H, m)], an oxymethine proton [$_{\rm H}$ 4.75 (1H, d, J = 5.5 Hz)], and a series of aliphatic multiplets (Table 1). The ¹³C NMR spectrum (Table 1), associated with DEPT NMR spectrum, displayed 20 distinct carbon resonances attributed to a ketone carbonyl (c 216.1), an ester carbonyl group (c 170.4), a terminal double bond (c 120.8 and 141.7), a trisubstituted double bond (_c 131.6 and 133.1), three methyl groups (c 14.2, 22.6, and 25.5), four methines (one oxygenated at c 81.9), five methylenes, and two quaternary carbons ($_{\rm C}$ 35.2 and 47.4). The above mentioned information was similar to that of a co-isolated ent-abietane diterpenoid 3 (isoforrethin C),³⁷ except for the major differences being due to the presence of an additional ketocarbonyl group (c 216.1, C-3) and the absence of the oxymethine carbon in 1, suggesting that 1 was an oxidized derivative of 3. The location of the ketocarbonyl group was assigned at C-3 according to the HMBC correlations from H₃-18 and H₃-19 to ketocarbonyl (Fig. 2). The detailed 2D NMR data analysis confirmed the planar structure of 1, which could be further confirmed through the oxidization of 3 with PCC in dichloromethane.

The relative configuration of 1 was established by NOESY experiments and ${}^{1}H - {}^{1}H$ coupling constants. The strong NOE corrections of H-5/Me-18 and Me-20/Me-19, along with the absence of NOE correlation between H-5 and Me-20 suggested the trans-relationship of H-5 and Me-20, and they occupied the axial bonds of the chair conformational cyclohexanone ring A. Thus, H-5 was arbitrarily designated as -orientation, while Me-20 was accordingly assigned as -orientation. Then, the NOE correlations of H-9/H-5 and H-12 assigned that H-9 was -oriented. Subsequently, the NOE correlations H-13/H-12 and H-13/H-14 assigned H-13 and H-14 to be -orientation. This assignment was also supported by the coupling constants of H-13 (ddd, J = 10.8, 5.8, 5.8 Hz) and H-14 (d, J = 5.5 Hz) on the chair conformational C ring. Moreover, the absolute configuration of 1 was established by comparing its experimental ECD spectra with those calculated by the time-dependent density functional theory (TDDFT). As shown in Fig. 3, the experimental ECD curve of 1 showed a negative Cotton e ect at 212 nm, indicating that **1** had a (5*S*,9*S*,10*S*,13*S*,14*R*)-configuration. Thus, compound 1 was established as depicted, and was given a trivial name isodopene A.

Isodopene B (13) was obtained as white powder. The HRESIMS ion at m/z 525.3560 [M + Na]⁺ (calcd for C₃₁H₅₀O₅Na, 525.3550) suggested a molecular formula $C_{31}H_{50}O_5$ for **13**. The IR spectrum showed absorption of hydroxyl group (3337 cm⁻¹) and carbonyl group (1736 cm⁻¹). The ¹H NMR data of **13** showed signals of seven methyl groups [H 0.78 (3H, s), H 0.80 (3H, s), 0.94 (3H, d, J = 6.6 Hz), 0.98 (3H, s), 1.12 (3H, s), 1.19 (3H, s), 1.23 (3H, s)], a methoxyl group [H 3.33 (3H, s), two oxygenated methylene protons [H 3.03 (1H, d, J = 12.8 Hz) and 3.78 (1H, d, J = 12.9 Hz)], an hemiacetal proton [H 4.57 (1H, dd, J = 9.4, 5.0 Hz)], an olefinic proton [$_{\rm H}$ 5.36 (1H, dd, J = 3.9, 3.9 Hz)] (Table 2). The ¹³C NMR spectrum (Table 2), associated with DEPT NMR spectrum, classified thirty one carbon resonances for a carboxyl group (c 184.0), a trisubstituted double bond (_c 129.9 and 137.9), a hemiketal carbon (_c 101.1), four methines, seven methyls, nine methylenes (one oxygenated at _c 70.6), an oxygenated tertiary carbon ($_{\rm C}$ 73.2), and five sp³ quaternary carbons. The abovementioned information showed high similarity with that of a coisolated 2,3-seco-triterpene 14 (maquatic acid),³⁸ except for the presence of an additional methoxyl group in 13, indicating that 13 was a methylated derivative of 14. HMBC correlation from H-2 to the methoxyl carbon (_C 54.6) located the methoxyl group at C-2. Further detailed analysis of the 2D NMR data confirmed the planar structure of 13

The stereochemistry of **13** was determined by analysis of its NOESY correlations. The NOESY correlations of H-9/H-5 and Me-26/Me-25 suggested that H-5, H-9, Me-26, and Me-25 occupied the axial bonds of the chair conformational B ring (Fig. 2). Thus, H-5 and H-9 were arbitrarily designated as -orientations, while Me-26 and Me-25 were accordingly assigned as -orientation. Subsequently, the NOE correlations Me-25/H-2 suggested that H-2 was -orientation. The NOE correlations H-9/Me-27, Me-27/H-16, and H-16/H-22 indicated that Me-27 was -orientation. Thus, on the chair conformational E ring, the NOE correlations of H-18/H-22, Me-29, and H-20 assigned H-18, Me-29, and H-20 to be -orientation.

The absolute configuration of **13** was established by comparing its experimental ECD spectra with those calculated by the time-dependent density functional theory (TDDFT). As shown in Fig. 3, the experimental ECD curve of **13** showed one positive (187 nm) and one negative (233 nm) Cotton e ects, respectively, indicating that **13** had a (2*S*,5*R*,8*R*,9*R*,10*R*,14*S*,17*S*,18*S*,19*R*,20*R*)-configuration. Thus, compound **13** was established as depicted, and was given a trivial name isodopene B.

The 25 known compounds were identified as isoforrethin D (2),³⁷ isoforrethin C (3),³⁷ macrophynin E (4),³⁸ 19-hydroxyferruginol (5),³⁹ (–)-lambertic acid (6),³⁸ 3 ,7 -dihydroxy-*ent*-kaur-16-en-15-one (7),⁴⁰ *ent*-3 -hydroxykaurene-15-one (8),⁴¹ inumakoic acid (9),⁴² trogopteroid F (10),⁴³ isoliquiritigenin (11),⁴⁴ echinatin (12),⁴⁵ maquatic acid (14),⁴⁶ ursolic acid lactone (15),⁴⁷ 3-oxo-urs-12-en-28-oic acid (16),⁴⁸ 2,4 ursolic acid acetate (17),⁴⁹ pomolic acid 3 -acetate (18),⁵⁰ 3 ,28-dihydroxyursulane (19),⁵¹ euscaphic (20),⁵² corosolic acid (21),⁵³ eucalyptolic acid (22),⁵⁴ epimaslinic acid (23),⁵⁵ 2 ,3 ,24-tri-hydroxyolean-12-en-28-oic acid (25),⁴¹ 2 ,3 ,23-trihydroxy-olean-12-en-28-oic acid (26),⁵⁷ and oleanolic acid (27),⁵⁸ by comparison of their NMR data with those in the literature.

2.2. TDP1 inhibition

The isolates were firstly screened for their TDP1 inhibition through a fluorescence assay at 100 μ M concentration. Twenty-one compounds **1–12**, **14**, **17–23**, and **25** showed TDP1 inhibition with the percentage inhibition ranging from 20% to 99%. These twenty-one active compounds were further tested for IC₅₀ values, defined as the compound concentration that results in 50% enzyme activity inhibition. As shown in Table 3, most compounds showed low TDP1 inhibitory activity with IC₅₀ values more than 100 μ M except two chalcone derivatives



Fig. 2. Key 2D NMR correlations of 1 and 13.

isoliquiritigenin (11) and echinatin (12) showing moderate inhibition with IC_{50} values of 55 μM and 45 μM , respectively.

2.3. TOP1 inhibition

TOP1 inhibitory activity of the isolates was assessed by TOP1mediated relaxation assay with CPT as a positive control, and semiquantitatively expressed relative to CPT at 100 μ M as follows: + + + +, more than 90% of the activity; + + +, between 60% and 89% of the activity; + +, between 30% and 59% of the activity; +, less than 29% of the activity; 0, no activity. As shown in Table 3, five compounds 8, **16**, **17**, **20**, and **22** exhibited equipotent inhibitory activity to CPT with TOP1 inhibition of + + + + (Fig. 4A). Compounds 8, **16**, and **22** showed well dose-dependent inhibition of TOP1 at the concentrations of 0.4, 2, 10, 50, and 250 μ M. Four compounds **1**, **7**, **24**, and **25** showed high TOP1 inhibition of + + +. Eleven compounds, including the new compound **13** and the most potent TDP1 inhibitors **11** and **12** showed moderate TOP1 inhibition (+ +).

To study the inhibitory mechanism of the active compounds, **8**, **16**, and **22** were selected for further studies. TOP1 poison CPT could trap and stabilize TOP1cc and was used as the positive control. TOP1-mediated DNA cleavage assay indicated that the nicked DNA content increased in the lanes with the presence of CPT (Fig. 4B). With the presence of **8**, **16**, and **22**, the nicked DNA content decreased, implying that these compounds could inhibit the formation of TOP1cc. TOP1-mediated EMSA assay (Fig. 4C) indicated that the tested compounds did not hamper the binding of TOP1 to DNA. The TOP1-mediated assays indicated that the compounds **8**, **16**, and **22** act as TOP1 catalytic inhibitors, which prevent the formation of TOP1-DNA covalent cleavage complex by blocking the nucleophilic attack of TOP1 to the scissile strand.



Fig. 3. Calculated and experimental ECD spectra of 1 and 13 in MeCN.

Table 2 $^1\mathrm{H}$ (500 MHz) and $^{13}\mathrm{C}$ (125 MHz) NMR data of 13 in CDCI_3

No.	_H (mult., <i>J</i> in Hz)	_c , type	No.	_H (mult., <i>J</i> in Hz)	_c , type
1	2.05, m	43.9, CH ₂	15	1.70, m	28.3, CH ₂
	1.39, m			1.04, m	
2	4.57, dd (9.4, 5.0)	101.1, CH	16	2.51, m	25.6, CH ₂
				1.60, m	
			17		48.0, C
3	3.78, d (12.9)	70.6, CH ₂	18	2.54, s	53.1, CH
	3.03, d (12.8)		19		73.2, C
4		39.2, C	20	1.41, m	41.2, CH
5	0.83, m	60.1, CH	21	1.68, m	26.1, CH ₂
				1.31, m	
6	1.47, m	20.9, CH ₂	22	1.80, m	37.5, CH ₂
	1.36, m			1.67, m	
7	1.46, m	33.0, CH ₂	23	0.80, s	28.0, CH ₃
	1.31, m				
8		40.7, C	24	0.98, s	21.1, CH ₃
9	1.59, m	44.2, CH	25	1.12, s	14.2, CH ₃
			26	0.78, s	17.6, CH ₃
10		40.8, C	27	1.23, s	24.4, CH ₃
11	2.12, m	24.8, CH ₂	28		184.0, C
	1.99, m				
12	5.36, dd (3.9, 3.9)	129.9, CH	29	1.19, s	27.5, CH ₃
13		137.9, C	30	0.94, d (6.6)	16.3, CH ₃
14		41.8, C	-OCH ₃	3.33, s	54.6, CH ₃

Table 3

The TDP1 and TOP1 inhibitory activities, and the cytotoxicity of the isolated compounds.

No.	Inhibition		Cytotoxicity (GI ₅₀ , µM) ^c			
	TDP1 ^a	TOP1 ^b	MCF-7	A549	HCT116	
1	> 100	+ + +	15 ± 5.6	51 ± 1.7	$31~\pm~0.4$	
2	No ^a	+ +	40 ± 1.4	51 ± 2.1	42 ± 3.2	
3	> 100	+ +	53 ± 5.1	35 ± 1.1	35 ± 3.2	
4	> 100	0	> 100	> 100	> 100	
5	> 100	+ +	92 ± 2.4	91 ± 3.0	51 ± 1.7	
6	> 100	+ +	60 ± 5.9	36 ± 2.2	68 ± 3.3	
7	> 100	+ + +	3.5 ± 0.4	4.8 ± 0.5	2.7 ± 0.3	
8	> 100	+ + + +	2.2 ± 1.1	3.0 ± 0.4	2.8 ± 0.4	
9	> 100	+ +	60 ± 2.2	41 ± 14	31 ± 7.1	
10	> 100	+ +	49 ± 2.3	24 ± 0.9	43 ± 3.1	
11	55 ± 0.4	+ +	> 100	87 ± 2.8	39 ± 4.3	
12	45 ± 0.5	+ +	43 ± 1.2	36 ± 1.2	64 ± 4.4	
13	> 100	+ +	31 ± 1.5	63 ± 2.5	37 ± 8.3	
14	> 100	+	> 100	> 100	23 ± 5.1	
15	No	0	> 100	> 100	> 100	
16	No	+ + + +	12 ± 0.9	57 ± 4.1	5.3 ± 0.8	
17	> 100	+ + + +	6.5 ± 2.0	48 ± 1.6	21 ± 3.7	
18	> 100	0	> 100	> 100	> 100	
19	> 100	+ +	93 ± 6.1	45 ± 2.4	37 ± 1.5	
20	> 100	+ + + +	41 ± 5.0	43 ± 8.4	22 ± 3.6	
21	> 100	+	62 ± 8.1	> 100	> 100	
22	> 100	+ + + +	67 ± 2.1	59 ± 3.7	22 ± 2.8	
23	> 100	+ +	> 100	> 100	51 ± 1.3	
24	No	+ + +	14 ± 0.02	22 ± 1.3	29 ± 2.0	
25	> 100	+ + +	11 ± 1.1	37 ± 2.8	55 ± 3.0	
26	No	0	> 100	> 100	> 100	
27	No	0	> 100	> 100	> 100	

 a TDP1 inhibition was determined by using a fluorescence assay and expressed as IC_{50} (\mu M) values. Every experiment was repeated at least three times independently.

 $^{\rm b}$ Top1 inhibitory activity was semiquantitatively expressed relative to CPT at 100 μM as follows: + + + +, > 90% of the activity; + + +, between 60% and 89% of the activity; + +, between 30% and 59% of the activity; +, less than 29% of the activity; 0, no activity.

^c GI₅₀ values were defined as the concentrations of compounds resulting in 50% cell growth inhibition. Every experiment was repeated at least three times independently.

^d No TDP1 inhibitory activity at 100 µM concentration.

2.4. Cytotoxicity

The cytotoxicity of the isolates was evaluated through MTT assay against three human cancer cell lines, including breast cancer (MCF-7), non-small cell lung cancer (A549), and colon cancer (HCT116) cell lines. The GI₅₀ values, defined as the concentrations resulting in 50% cell growth inhibition, are summarized in Table 3. The potent TOP1 inhibitors, *ent*-kaurane diterpenoids **7** (+ + +) and **8** (+ + + +) displayed significant cytotoxicity against three human cancer cells with GI₅₀ values ranging from 2.2 to 4.8 μ M. Compound **16** showed high cytotoxicity against HCT116 (GI₅₀ = 5.3 μ M) with TOP1 inhibition of + + + +. Compound **17** showed high cytotoxicity against MCF-7 cells (GI₅₀ = 6.5 μ M) with TOP1 inhibition of + + + +. Although compounds **20** and **22** had potent TOP1 inhibition (+ + + +), they showed moderate cytotoxicity against these three cancer cells.

2.5. Synergistic e ect

Because the chalcone derivatives **11** and **12** showed moderate TDP1 inhibition, their synergistic activity with the TOP1 poison topotecan (TPT) was evaluated in MCF-7 cells by using MTT assay. After being incubated for 72 h at 37 °C, the cytotoxicity of TPT against MCF-7 Cells significantly increased in the presence of both **11** and **12** (Fig. 5, left). The combination analyses of the does-response data were presented in form of Combination Index (CI) vs. Fraction A eted (Fa) plots (Fig. 5, right), demonstrating that **11** showed synergistic e ect (CI values < 1) and **12** showed strong synergistic e ect with TPT (CI values < 0.3) in MCF-7 cells.

2.6. DNA damage

To assess the DNA damage e ect, H2AX foci formation induced by **8** and **16** was conducted by immunofluorescence microscopy in breast cancer MCF-7 cells. After being incubated with **8** and **16** for 12 h, MCF-7 cells were stained with H2AX antibodies. As shown in Fig. 6, positive control CPT could e ectively induce H2AX foci formation at 0.5 μ M concentration. On the contrary, there were only weak DNA damage observed after being incubated with **8** or **16** up to 9 μ M concentration, implying that both **8** and **16** could not induce TOP1-mediated DNA break which is consistent with their molecular action for TOP1.

3. Conclusion

In summary, twenty-seven compounds, including a new ent-abietane diterpenoid (1) and a new 2,3-seco-triterpene (13) were isolated from the EtOH extract of the roots of I. ternifolius. The two new compounds 1 and 13 showed strong and moderate TOP1 inhibition of + + + and + +, respectively. To the best of our knowledge that two chalcone derivatives 11 and 12 were firstly found as dual TDP1 and TOP1 natural inhibitors, and showed synergistic e ect with TPT in MCF-7 cells. Five compounds 8, 16, 17, 20, and 22 showed strong TOP1 inhibitory activity of + + + +. Further studies indicated that compounds 8, 16, and 22 could prevent the formation of covalent cleavage complex acting as TOP1 catalytic inhibitors, and antagonize TOP1-mediated DNA break in MCF-7 cells. Compound 8 exhibited the highest cytotoxicity against MCF-7 and A549 cells with GI₅₀ value of 2.2 and 3.0 µM, respectively. Compound 7 showed the highest cytotoxicity against HCT116 cells with GI₅₀ value of 2.7 µM. These results imply that I. ternifolius have multiple health functions, and will attract more attention for food scientists because it is consumed as cattle feed and has nutritional and pharmaceutical values.



Fig. 4. (A) TOP1-mediated relaxation assays. Lane 1, pBR322 DNA alone; Iane 2, pBR322 DNA and TOP1; Ianes 3–7, pBR322 DNA, TOP1 and CPT at 0.2, 1, 5, 25, 125 µM, or the tested compounds at 0.4, 2, 10, 50, 250 µM concentrations, respectively. (B) TOP1-mediated DNA cleavage assays. Lane 1 pBR322 DNA alone; Iane 2, pBR322 DNA and TOP1; Ianes 3–10, pBR322 DNA, TOP1 and the drugs at 25 and 50 µM concentrations, respectively. (C) DNA mobility shift analysis. Lane 1, pBR322 DNA alone; Iane 2, pBR322 DNA and TOP1; Ianes 3–10, pBR322 DNA, TOP1 and the drugs at 25 and 50 µM concentrations, respectively. (C) DNA mobility shift analysis. Lane 1, pBR322 DNA alone; Iane 2, pBR322 DNA and TOP1; Ianes 3–10, pBR322 DNA, TOP1 and the drugs at 25 and 50 µM concentrations, respectively. R, relaxed DNA; Sc, supercoiled DNA; N, nicked DNA; C, TOP1-bound DNA.

4. Materials and methods

4.1. General experimental procedures

Optical rotations were measured on a Perkin-Elmer 341 polarimeter (PerkinElmer, Inc. Shelton, USA). Melting points were measured on an X-5 melting instrument without being corrected. UV spectra were recorded on a Shimadzu UV-2450 spectrophotometer. IR spectra were determined on a Bruker Tensor 37 infrared spectrophotometer with KBr pellets. CD spectra were obtained on an Applied Photophysics Chirascan spectrometer. HRESIMS were measured on a Shimadzu LCMS-IT-TOF spectrometer. Nuclear magnetic resonance spectra were measured on a Bruker AM-500 spectrometer using tetramethylsilane as an internal reference. A SEP LC-52 equipped with UV-200 and a Dr. Maisch C₁₈ column (250 \times 10 mm, S-5 μ m) was used for semi-preparative HPLC separation. RP-C₁₈ silica gel (YMC, 50 µm), MCI gel (CHP20P, 75–150 µm, Mitsubishi Chemical Corporation, Tokyo, Japan), silica gel (100-200 and 200-300 Mesh Marine Chemical Ltd, Qingdao, People's Republic of China), and sephadex LH-20 (GE Healthcare Bio-Sciences AB, Sweden) were used for column chromatography (CC). TLC was carried out on silica gel GF₂₅₄ plates (Qingdao Marine Chemical, Inc. China). All solvents were analytical grade from the local suppliers without further purification. TDP1 protein was expressed and purified from Escherichia coil BL21 (DE3) cells in our lab. Plasmid pBR322 DNA and calf thymus DNA TOP1 were purchased from Takara Biotechnology (Dalian, People's Republic of China). A549, HCT116 and MCF-7 cell lines were purchased from China Center for Type Culture Collection.

4.2. Plant material

The roots of *I. ternifolius* were collected in June from Yulin county, Guangxi province, People's Republic of China, and identified by Dr. Gui-Hua Tang, Sun Yat-sen University. A voucher specimen (No. 2018121101) was deposited at the School of Pharmaceutical Sciences, Sun Yat-sen University.

4.3. Extraction and isolation

The air-dried powder of the roots of I. ternifoliu (20 kg) was extracted with 95% EtOH (20 L \times 3) at room temperature to give a crude extract (3 kg). The crude extract was suspended in H₂O (2.5 L) and partitioned with EtOAc (5 L \times 3). The EtOAc fraction (506 g) was subjected to silica gel CC eluted with a petroleum ether/EtOAc gradient (100:1 to 0:1, v/v) to obtain ten fractions (Fr.A – Fr.J). Fr.I was separated by RP-18 silica gel CC, eluted with MeOH/H₂O gradient (30% to 100%, v/v) to give four subfractions (Fr.Ia - Fr.Id). Fr.Id was separated by silica gel CC eluted with a CH₂Cl₂/MeOH gradient (100:0 to 100:20, v/v) to obtain five fractions (Fr.Id1 - Fr.Id5). Fr.Id1 was separated by sephadex LH-20 and followed by semi-preparative HPLC (75% MeCN in H₂O, 3.0 mL/min) to give **2** (8.8 mg, $t_{\rm R}$ = 13.5 min), **3** (7 mg, $t_{\rm R}$ = 21 min). Fr. Id2 was separated by sephadex LH-20 and followed by semi-preparative HPLC (60% MeCN in H₂O, 3.0 mL/min) to give **1** (6 mg, $t_{\rm R}$ = 16 min). Fr.Id3 was separated by silica gel CC (CH₂Cl₂/MeOH, 125:1), and followed by sephadex LH-20 (MeOH) to give 22 (70 mg). Fr.Id4 was separated by silica gel CC (CH₂Cl₂/MeOH, 50:1), sephadex LH-20 (MeOH), and followed by preparative TLC





Fig. 5. Potentiation of the cytotoxic action of TPT in combination with 11 (A) and 12 (B) in MCF-7 cells. Left panels: dose response curves for combination treatment with TPT and 11 or 12. Right panels: Combination Index (CI) vs. Fraction A ected (Fa) plot for the dose response graphs in the left panels. CI < 1, synergism; CI < 0.3, strong synergism.

(CH₂Cl₂/MeOH, 70:1) to give 20 (20 mg), 24 (7.8 mg), and 26 (8 mg). Fr.Id5 was separated by sephadex LH-20 and further purified by silica gel CC (CH₂Cl₂/MeOH, 130:1, v/v) to give 13 (10 mg) and 15 (11.8 mg). Fr.Ic was subjected to silica gel CC (CH₂Cl₂/MeOH, 100:1 to 5:1, v/v) to give six fractions (Fr.Ic1 - Fr.Ic6). Fr.Ic2 was separated by sephadex LH-20 (MeOH), and followed by silica gel CC (CH₂Cl₂/MeOH, 1:0 to 100:1, v/v) to give 16 (20 mg), 17 (50 mg), and 27 (200 mg). Fr.Ic3 was separated by silica gel CC (CH₂Cl₂/MeOH, 100:1, v/v) to give 23 (14 mg), 25 (30 mg), 18 (10 mg), 19 (7 mg), 21 (8.6 mg), and 14 (18 mg). Fr.Ic4 was separated by semi-preparative HPLC (70% MeCN in H₂O, 3.0 mL/min) to give **11** (6 mg, $t_{\rm R}$ = 12 min). Fr.Ic5 was separated by semi-preparative HPLC (75% MeCN in H₂O, 3.0 mL/min) to give **12** (3 mg, $t_{\rm R}$ = 15 min). Fr.H was separated by MCI gel CC (MeOH/H₂O, 50:50 to 100:0, v/v) to obtain three fractions (Fr.H1 – Fr.H3). Fr.H2 was separated by silica gel CC (CH₂Cl₂) to give two subfractions (Fr.H2a - H2bFr.). Fr.H2a was separated by silica gel CC (petroleum ether/CH₂Cl₂, 1:1, v/v), and followed by sephadex LH-20 (MeOH) to give 4 (12 mg) and 5 (9 mg). Fr.H2b was separated by sephadex LH-20 (MeOH), and followed by silica gel CC (petroleum ether/CH₂Cl₂, 3:1, v/v) to give 6 (18 mg), 9 (7 mg), and 10 (3.8 mg). Fr.H3 was separated by sephadex LH-20 (MeOH) and RP-18 silica gel CC (83% MeOH in H_2O , v/v) to give **8** (9 mg) and **7** (4.8 mg).

4.3.1. Isodopene a (1)

White powder; mp 159.7–161.3 °C; [] – 32.3 (c 0.3, MeCN); UV (MeCN) max (log ϵ) 217 (3.20) nm; ECD (MeCN) 212 (ϵ – 6.94) nm; IR

(KBr) ν_{max} : 3374, 2919, 2852, 1762, 1707, and 1668 cm $^{-1}$. 1H NMR (500 MHz, CDCI₃) and ^{13}C NMR (125 MHz, CDCI₃) date see Table 1. HRESIMS m/z 337.1771 [M + Na] $^+$ (calcd for $C_{20}H_{26}O_3Na^+$, 337.1774).

4.3.2. Isodopene B (13)

White powder; mp 182.9–184.6 °C; [] + 65 (*c* 0.3, MeCN); UV (MeCN) $_{max}$ (log ε) 227 (3.10) nm; ECD (MeCN) 187 (ε + 18.31), 223 (ε – 2.24) nm; IR (KBr) ν_{max} : 3337, 2954, 2921, 2852, and 1736 cm⁻¹. ¹H NMR (500 MHz, CDCI₃) and ¹³C NMR (125 MHz, CDCI₃) date see Table 2. HRESIMS *m*/*z* 525.3560 [M + Na]⁺ (calcd for C₃₁H₅₀O₅Na⁺, 525.3550).

4.4. TOP1-mediated relaxation assay

All isolates were tested for the TOP1 inhibitory activity using TOP1mediated relaxation assay.³¹ Briefly, the reaction mixture (20 μ L) containing 1 μ L of supercoiled pBR322 DNA (0.5 μ g/ μ L) in Relaxation Bu er (10 mM Tris-HCl, pH 7.5, 50 mM KCl, 5 mM MgCl₂, 15 μ g/mL BSA, 40 μ g/mL DTT) was incubated with 1 unit of calf thymus TOP1 in the absence or presence of the tested compounds for 30 min at 37 °C. Then the reaction solution was added 6 × loading bu er (4 μ L), and was analyzed using a 0.8% agarose gel in TBE bu er at 4.6 V/cm for 1.5 h. The gel was stained with 1 × gel red for 30 min and subsequently visualized with a UV transilluminator.



Fig. 6. Fluorescence microscopy detection of H2AX foci formation in MCF-7 cells. Cells were treated with CPT (0.5 µM) or the compounds 8 and 16 (9 µM) for 12 h.

4.5. TOP1-mediated cleavage assay

TOP1-mediated cleavage assay was carried out as mentioned before.³² The reaction mixture (20 μ L) containing 1 μ L of supercoiled pBR322 DNA (0.5 μ g/ μ L) and 10 units of calf thymus TOP1 in Relaxation Bu er was incubated for 30 min at 37 °C. Then, the reaction solution was added with 4 μ L of proteinase K (5.5 mg/mL) and incubated at 55 °C for 15 min. The reaction was terminated by the addition of 5 μ L 6 \times loading bu er and was analyzed using a 0.8% agarose gel in TBE bu er at 3 V/cm for 30 min. The gel was stained with ethidium bromide (EB, 0.125 μ g/mL) in TBE bu er for 30 min. Finally, the gel was run in TBE bu er for another 30 min and was visualized with a UV transilluminator.

4.6. TOP1-mediated EMSA

TOP1-mediated DNA mobility shift analysis (EMSA) assay was used to evaluate whether the compounds hampered the binding of TOP1 to DNA.³³ Supercoiled pBR322 DNA (1 μ L, 0.1 μ g/ μ L) in 20 μ L of Relaxation Bu er was incubated with 4 units of TOP1 in the absence or presence of the tested compounds at 37 °C for 0.5 h. Then, the reaction solution was added with 4 μ L 6 × loading bu er and was analyzed using a 0.8% agarose gel with 1% EB in TBE bu er at 0.8 V/cm for 6.5 h. The gel was visualized with a UV transilluminator.

4.7. TDP1 inhibition assay

The TDP1 fluorescence assay was conducted according to the reported method.³⁴ Briefly, the reaction mixture (50 µL) containing 20 µL of TDP1 solution [0.02 µL of purified TDP1 (100 nM) in 10 mM Tris-HCl, pH 7.5, 50 mM KCl, 1 mM EDTA, 1 mM DTT] was dispensed into a white 384-well plate. The tested compound solution in DMSO (5 µL) was pinned into assay plates. The solution was incubated at 37 °C for 30 min. Then, the plate was read by molecular devices at Ex_{485}/Em_{510} nm to identify false-positive compounds that had autofluorescence. The linear oligonucleotide substrate (5 -*FAM*-AGGATCTAAAAGACTT-BHQ-

3 , 25 μ L, 35 nM) was dispensed into the wells to start the reaction. The plate was immediately read three times at Ex_{485}/Em_{510} nm. TDP1 inhibition percentage was calculated by comparing the rate of enhancement in fluorescence intensity for the compound-treated wells to that of DMSO control wells.

4.8. Cell culture and cytotoxicity assays

The cells were cultured on DMEM or RPMI-1640 medium at 37 °C in a humidified atmosphere with 5% CO₂. The cytotoxicity of compounds **1–27** was evaluated through MTT assay against three human cancer cell lines (breast cancer MCF-7, non-small cell lung cancer A549, and colon cancer HCT116). The cancer cells were treated with the compounds (predissolved in DMSO) at a five-dose assay ranging from 0.01 to 100 μ M concentration for 72 h. MTT solution (20 μ L, 2.5 mg/mL) in PBS bu er was added. After being incubated for 4 h, the formazan crystal formed in the well was dissolved in 100 μ L of DMSO for optical density at 570 nm. The Gl₅₀ value was calculated by nonlinear regression analysis (GraphPad Prism). Every experiment was conducted for three independent times.

4.9. Synergistic e ect

The drug combination experiments were measured by through MTT assay.³⁵ MCF-7 cells were treated with the tested compounds and TOP1 inhibitor topotecan for 72 h at 37 °C, and then measured by MTT assay. The combination index (CI) values were calculated by chou-talalay method (calcusyn software).

4.10. H₂AX detection

H2AX staining was performed according to the reported method.³⁶ MCF-7 cells (5 × 10⁴ cells/mL) were grown in culture medium and treated with the tested compounds for 12 h. After incubation, cells were fixed in 4% paraformaldehyde/PBS for 15 min at 25 °C, then permeabilized with 0.5% triton-X100/PBS at 37 °C for 30 min, and finally blocked with 5% goat serum/PBS at 37 °C for 3 h. Immunofluorescence assay was performed using standard methods, and the slides were incubated alternately with phospho- H2AX (Cell Signaling Technology) at 37 °C overnight. The cover slips were washed six times with blocking bu er and then incubated with anti-rabbit alexa 488-conjugated antibody (A21206, Life Technology) at 37 °C for 2 h. The dishes were again washed six times with blocking bu er. Digital images were recorded using a FLUOVIEW FV3000 (Olympus, Japan) and analyzed with FV31S-SW software.

Declaration of Competing Interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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Appendix A. Supplementary material

IR, CD, HRESIMS, and NMR (1 H NMR, 13 C NMR, DEPT135, 1 H- 1 H COSY, HSQC, HMBC and NOESY) spectra of Isodopene A (1) and B (13).

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